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(2/3)

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Go From Love to
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How To Speak by
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46 Minutes | John
Assaraf on
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Davis [NEW]**

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the Environment,

and the Universe 3

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Bonds Use with

Chapter 7, Section

7.2 1. How many

valence electrons

does a neutral

magnesium (Mg)

atom have? 2.

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What is the charge on a magnesium ion? What does magnesium have to do to form such an ion, and why does it tend to do so? 3. How many valence electrons does a single neutral chlorine (Cl) atom have? 4.

Snow Elementary

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WORKSHEET The

Activity Series Use

with Chapter 9,

Section 9.2 1. For

each of the

following pairs of

elements,

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Underline the one
that would replace
the other element
in a compound.

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Atomic Orbitals 1.
What is the shape
of an s orbital?

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Chapter 5, Section

5.2 2. What is the relationship between the size of

an s orbital and the principal energy level in which it is found? 3. What is the shape of a p orbital? How many p orbitals are there in a level? 4.

Bryant Middle

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Chemistry

Formation Of Ions

Master 2
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Chapter 9, be sure
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Chemistry: Matter

and Change 3

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Formation of Ions

1. calcium and oxygen
2. Yes; each contains equal numbers of protons and electrons.
3. ionization energy
4. positive; cation
5. negative; anion
6. No; it is not a

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stable octet of
electrons. 7.

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