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39 E On this test, the easiest way to balance an equation is by plugging in the answers. The test writer tells you that the coefficient in front of H_2O is 6, so put that in there. Now, see which of the answer choices, if placed in front of NF_3 , would result

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in a balanced equation. If, for instance, we try choice (B), we'd have 1 mole of N on the left, which would give us 3 moles of F on the left.

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Answers 1. $1 \text{ SnO}_2 + 2 \text{ H}_2 \rightarrow 1 \text{ Sn} + 2 \text{ H}_2\text{O}$ 2. $3 \text{ KOH} + 1 \text{ H}_3\text{PO}_4 \rightarrow 1 \text{ K}_3\text{PO}_4 + 3 \text{ H}_2\text{O}$ 3. $2 \text{ KNO}_3 + 1 \text{ H}_2\text{CO}_3 \rightarrow 1 \text{ K}_2\text{CO}_3 + 2 \text{ HNO}_3$ 4. $1 \text{ Na}_3\text{PO}_4 + 3 \text{ HCl} \rightarrow 3 \text{ NaCl} + 1 \text{ H}_3\text{PO}_4$ 5. $1 \text{ TiCl}_4 + 2 \text{ H}_2\text{O} \rightarrow 1 \text{ TiO}_2 + 4 \text{ HCl}$ 6. $1 \text{ C}_2\text{H}_6\text{O} + 3 \text{ O}_2 \rightarrow 2 \text{ CO}_2 + 3 \text{ H}_2\text{O}$ 7. $2 \text{ Fe} + 6 \text{ HC}_2\text{H}_3\text{O}_2 \rightarrow 2 \text{ Fe}(\text{C}_2\text{H}_3\text{O}_2)_3 + 3 \text{ H}_2$ 8. $4 \text{ NH}_3 + 5 \text{ O}_2 \rightarrow 4 \text{ NO}$

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+ 6 H₂O 9. 1 B₂Br₆ + 6
HNO₃ ? 2 B(NO₃)₃ + 6 HBr
10. 4 NH₄OH + 1 K₂(SO₄)
2 · 12H₂O ? 1 Al(OH)₃ + 2
(NH₄)₂SO₄ ...

Balancing Equations

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Chemistry is a very fun subject and being a 10th grader it is important for you to ensure that you understand each topic taught before the final quiz. The quiz below is designed to

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see just how attentive you were in class this past semester, give it a shot and see just how much you remember so far. All the best!

Quiz: Chemistry For Grade 10 Student - ProProfs Quiz

This document gives detailed answer explanations to chemistry practice questions from The SAT Subject Tests™ Student Guide. By reviewing them, you'll get to know the types of questions on the test and learn your strengths and weaknesses. Estimated difficulty level is based on a 1-5 scale, with 1 the easiest and 5 the most difficult. For more

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about the SAT Subject Tests,
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1.

SAT Subject Tests Answer Explanations Chemistry

Chemistry 101: General
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