

Chapter 9 Test Chemistry

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,/Chemistry Chapter 9 (part-2) | Class 12 || Class 12 Chemistry Chapter 9 Chapter 9 Test

Chemistry

Chemistry Chapter 9 test 1. Write the names of the elements in the order listed in the formula. 2. Use prefixes appropriately to indicate the number of each kind of atom. 3. End the name of the second element with suffix -ide

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Chapter 9 - Ionic and Covalent Bonding . 1. In which pair do both compounds exhibit predominantly ionic bonding? A) RbCl and CaO B) PCl₅ and HF C) KI and O₃ D) Na₂SO₃ and BH₃ E) NaF and H₂O . ANS: A PTS: 1 DIF: easy REF: 9.1 OBJ: Define ionic bond. TOP: bonding | ionic bonding MSC: general chemistry . 2.

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Chapter 9 Practice Test - Naming and Writing Chemical Formulas Answer Section MATCHING 1. ANS: A PTS: 1 DIF: L1 REF: p. 253 OBJ: 9.1.1 Identify the charges of monatomic ions by using the periodic table, and name the ions. STA: Ch.5.a 2.ANS: H PTS: 1 DIF: L1 REF: p. 254

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PRENTICE HALL CHAPTER 9 CHEMISTRY TEST. 5. The metals in Group 1A, 2A and 3A: Circle Answer gain electrons when they form ions

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Table; important equation, constant, and data tables; and a glossary of chemistry terms Both teachers and test-taking students have praised earlier editions of this manual for its wealth of well-organized detail. Subject reviewed include the basics—matter, energy, scientific method, and measurements; atomic structure and the periodic table; bonding; chemical formulas; gases and laws; stoichiometry; liquids, solids, and phase changes; chemical reactions and thermochemistry; chemical reactions; chemical equilibrium; acids, bases, and salts; oxidation-reduction; carbon and organic chemistry; and the laboratory. **ONLINE PRACTICE TESTS:** Students who purchase this book or package will also get access to two additional full-length online SAT Chemistry subject tests with all questions answered and explained.

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