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(7.1) Ions Key Term. To find the  
number of valence electrons in an  
atom of a representative element,  
simply look at its group number. (7.1)  
Ions Key Term cont. A positively  
charged ion, or cation, is produced  
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valence electrons. (7.1) Ions Key Term  
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Chapter 7 Study Guide: Chemistry. To find the number of valence electrons in an atom of a representative element,

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Simply look at its group number. A positively charged ion, or cation, is produced when an atom loses one or more valence electrons.

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STUDY. PLAY. valence electrons.  
electrons in the highest occupied  
energy level of an element's atoms.  
Group Number, 1A= 1, 2A=2, 3A=3,  
4A=4, 5A=5, 6A=6, 7A=7, 8A=8. How  
can you determine the number of  
valence electrons an atom has from its  
position on the periodic table? Name  
by that

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~~Anion~~ negative charge. The charge for an Anion is written as an number followed by a minus sign. Cation. An ion with a positive charge. The charge for a cation is written as a number followed by a plus sign. Chemical formula.

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chemistry study guide chapter 7. what follows an element's symbol in a chemical formula to indicate the number of atoms of that element in one molecule of the compound? what does the chemical formula for an ionic compound represent?

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Organic compounds. Carbon compounds controls what happens in the cell of living organism and play key roles in the chemistry of living organisms. Why is lead no longer used in paints for plumbing pipes.

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Define chemical bond. Explain formation of cations and anions. Draw electron dot diagrams. Write electron configurations for ions. The force that holds two atoms together. Cations form when an atom loses electrons. An anion forms when.... Ex.  $K \rightarrow 1s^2 2s^2 2p^6 3s^2 3p^6 4s^1$ .

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dot structure. Octet rule. Halide ion.  
the electrons in the highest occupied  
energy level of an eleme.... diagrams  
that show valence electrons in the  
atoms of an elemen.... in forming  
compounds, atoms tend to achieve the  
electron confi....

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CHEM 109 Chapter 7: Energy Storage

either decreasing the hydrogen

pressure or increasing the

temperature, the H<sub>2</sub> can be re-

released on demand. Metal-hydride

storage systems are ideally suited for

PEM fuel cells that require high-purity

hydrogen. Because metal hydrides are

selective and absorb only hydrogen

and not larger gas molecules, they act

simultaneously ...

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force that holds two atoms together;

may form by the attraction of a positive

ion for a negative ion or by sharing

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**Answer:** a three-dimensional geometric arrangement of particles in which each positive ion is surrounded by negative ions and each negative ion is surrounded by positive ions; vary in shape due to sizes and relative numbers of the ions bonded.

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STUDY GUIDE FOR CHAPTERS 6 & 7. Certain topics like the structure, synthesis, and nomenclature of alkyl halides have already been covered previously. This part of the course is primarily concerned with  $S_N1$  and  $S_N2$  reactions,  $E1$  and  $E2$  reactions, and alkene synthesis, structure, and nomenclature. The following is the suggested order of readings and practice problems.

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